

**POLYSULFONE MIXED MATRIX MEMBRANES WITH BIOSYNTHESIS
NANOPARTICLES: ENHANCEMENT OF INTERFACE COMPATIBILITY
AND ANTIBACTERIAL MEMBRANE SEPERATION PROCESS**

KHAIRUL NAZRI BIN YUSOF

A thesis submitted in
fulfillment of the requirement for award of the
Doctor of Philosophy

Faculty of Mechanical and Manufacturing Engineering
Universiti Tun Hussein Onn Malaysia

AUGUST 2020

For my beloved mother and father



ACKNOWLEDGEMENT

Praise and thank to Allah for his guidance in helping me through my journey in preparing this thesis. It was through Him that I gained my strength to continue even though the road may seem tough and winding.

I would like to express my sincere gratitude and appreciation to my supervisors, Assoc. Prof. Dr. Zawati Harun, Assoc. Prof. Dr. Hatijah Basri and Dr. Zaini Yunos for their continuous support, help and guidance from the gathering of ideas until the completion of the report writing.

Special thanks to my beloved wife and family who have given me plenty of moral support, encouragements, motivation, sacrifice and love during the course of this project.

Finally, I would like to thank all my friends in AMMC who aided me in this research project



ABSTRACT

This study investigates the effect of biosynthesis silver nanoparticles (bio-AgNPs) structure towards PSf membranes performance. The addition of bio-AgNPs in the membrane formulation was aimed to improve antibacterial properties and interface compatibility of the polymer mixed matrix membrane. In this work, facile green synthesis method of silver nanoparticles was prepared using *Parkia speciosa* (Petai) leaves extract silver nitrate aqueous solution. The bio-AgNPs/PSf membranes was fabricated using phase-inversion process. Characterization of the synthesized bio-AgNPs and mixed matrix membranes via UV-Vis spectroscopy (UV-Vis), Field emission scanning electron microscope (FESEM), Transmission electron microscopy (TEM), X-ray diffraction (XRD), and Fourier transform infrared spectroscopy (FTIR). Surface plasmon resonance for bio-AgNPs was assigned at 465 nm with brown colour. FTIR spectroscopy identified the biomolecules capped on the surface of nanoparticles are phenol, flavonoid and terpenoid compounds. The microstructure and structural analyses had shown that AgNPs possessed good characteristics with spherical shapes, small average size of particles (59.96 nm), and small crystallite size. The bio-AgNPs also showed significant potential antibacterial activity against *Escherichia coli* (*E. coli*) and *Staphylococcus aureus* (*S. aureus*). The addition of bio-AgNPs was able to enhance the hydrophilicity of composite PSf membranes which proved by decreasing the value of contact angle. Meanwhile, structure on bottom layer shown a porous bulk with finger-like structure and macrovoid structure which is responsible for mechanical support. Finger-like structure on top layer became smaller with increment of bio-AgNPs because of its hydrophilicity properties. Hence, the pure water flux also will increase because the hydrophilicity properties was an essential factor for water permeability. The molecular interaction between molecules was investigated using Dynamic mechanical analysis (DMA) and silver leaching analysis by Inductive coupled plasma mass spectrometer (ICP-MS).

As revealed, the strength of molecular interaction between AgNPs and PSf molecules was improved with addition of bio-AgNPs. Hence, overall the results showed that the incorporation of bio-AgNPs able to improve antibacterial properties and interface compatibility of the polymer mixed matrix membrane.



ABSTRAK

Kajian ini menyiasat kesan biosintesis perak nanopartikel perak (bio-AgNPs) ke arah prestasi membran PSF. Penambahan bio-AgNPs dalam pembentukan membran bertujuan meningkatkan sifat antibakteria dan keserasian antara muka membran matrik campuran polimer. Dalam kerja ini, kaedah sintetik hijau mudah nanopartikel perak telah disediakan menggunakan ekstrak daun *Parkia speciosa* (Petai) dan larutan perak nitrat. Membran bio-AgNPs / PSf dihasilkan menggunakan proses fasa penyongsangan. Pencirian bio-AgNPs dan membran matriks campuran melalui spektroskopi UV-Vis (UV-Vis), mikroskop elektron pengimbasan pancaran (FESEM), mikroskop elektron penghantaran (TEM), X-ray difraksi sinar XRD, spektroskopi (FTIR). Resonans plasmon permukaan untuk bio-AgNPs adalah pada 465 nm dengan warna coklat. Spektroskopi FTIR mengenal pasti biomolekul yang dihidkan pada permukaan nanopartikel ialah sebatian fenol, flavonoid dan terpenoid. Analisis mikrostruktur dan struktur telah menunjukkan bahawa AgNP mempunyai ciri-ciri yang baik dengan bentuk bulat, saiz purata zarah kecil (59.96 nm), dan saiz kristal kecil. Bio-AgNP juga menunjukkan potensi aktiviti antibakteria yang signifikan terhadap *Escherichia coli* (E. coli) dan *Staphylococcus aureus* (S. aureus). Penambahan bio-AgNPs dapat meningkatkan hidrofilik membran PSF komposit yang terbukti dengan mengurangkan nilai sudut sentuhan. Sementara itu, struktur pada lapisan bawah menunjukkan liang berpori dengan struktur seperti jari dan struktur macrovoid yang bertanggungjawab untuk sokongan mekanikal. Struktur seperti jari pada lapisan atas menjadi lebih kecil dengan kenaikan bio-AgNPs kerana sifat hidrofiliknya. Oleh itu, fluks air tulen juga akan meningkat kerana sifat hidrofilik adalah faktor penting untuk kebolehtelapan air. Interaksi molekul antara molekul diselidiki menggunakan Analisis mekanik Dinamik (DMA) dan analisis peleburan perak oleh spektrometer massa plasma bersama Induktif (ICP-MS). Seperti yang dinyatakan, kekuatan interaksi molekul antara molekul AgNPs dan PSF

telah diperbaiki dengan penambahan bio-AgNPs. Oleh itu, secara keseluruhan keputusan menunjukkan bahawa pemerbadanan bio-AgNPs dapat memperbaiki sifat-sifat antibakteria dan keserasian antara muka membran matrik campuran polimer.



TABLE OF CONTENTS

	TITLE	i
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vii
	TABLE OF CONTENTS	ix
	LIST OF TABLES	xi
	LIST OF FIGURES	xii
	LIST OF ABBREVIATIONS	xvi
CHAPTER 1	INTRODUCTION	1
	1.1 Research background	1
	1.2 Problem statement	2
	1.3 Research objective	4
	1.4 Research scope	4
	1.5 Significant and novelty of research	5
CHAPTER 2	LITERATURE REVIEW	7
	2.1 Introduction to membrane technology	7
	2.2 Ultrafiltration membrane: polymer and ceramic	10
	2.3 Fouling challenge	22
	2.4 Biosynthesis of additive nanoparticles: AgNPs	24
	2.5 Antibacterial activity studies	28
	2.6 Toxicity of nanoparticles	30
	2.7 Summary	30

CHAPTER 3	METHODOLOGY	32
	3.1 Research design	32
	3.2 Silver nanoparticles (AgNPs)	34
	3.3 Membrane materials	38
	3.4 Membrane fabrication	39
	3.5 Membrane characterization	40
	3.6 Membrane performance	45
	3.7 Summary	50
CHAPTER 4	RESULTS AND DISCUSSIONS	51
	4.1 Synthesis of AgNPs	51
	4.2 Toxicity test	68
	4.3 RSM optimization	69
	4.4 Membrane characterization	73
	4.5 Water permeability	93
	4.6 Bonding of AgNPs	96
	4.7 Antibacterial activity	102
	4.8 Fouling test	105
CHAPTER 5	CONCLUSION	113
	REFERENCES	115
	APPENDIX	129



PT TAAUTHM
PERPUSTAKAAN TUNKU TUN AMINAH

LIST OF TABLES

2.1	The characteristics of nanoparticles for polymeric membranes	15
2.2	Review on development of UF PSf membrane (Yunos, 2014)	17
2.3	Some of the interface study in composite structure.	18
2.4	The effect of nanoparticles towards human health	19
2.5	Overview of types of nanomaterials applied for water and wastewater technologies	25
3.1	The model of CCD experimental	37
3.2	Content of materials used for membrane fabrication	40
4.1	Zone of inhibition ring for <i>S. aureus</i> and <i>E. coli</i> for bio-AgNPs	67
4.2	Yield of bio-AgNPs for CCD experimental	70
4.3	ANOVA for response surface linear model	71
4.4	The XRD results for 1% bio-AgNPs/PSf membrane	78
4.5	Contact angle value for modified PSf membranes	81
4.6	Mass of Ag element in membrane	86
4.7	Mean pore size for bio-AgNPs/PSf and AgNPs/PSf membranes	92
4.8	Water flux rate and humic acid rejection for bio-AgNPs/PSf membrane	96
4.9	Zone of inhibition ring for mixed matrix membranes	103
4.10	Fluxes of the membrane during UTHM lake water filtration process	112

LIST OF FIGURES

2.1	Basic principle of membrane separation.	8
2.2	Types of pressure driven membranes	9
2.3	Types of membranes (Yunos, 2014)	11
2.4	The cross section image of hybrid, composite and mixed matrix membranes.	12
2.5	The cross section structure for modified PSf membranes (Harun et al., 2016)	14
2.6	Phase inversion method	20
2.7	Ternary system with liquid-liquid demixing gap	21
2.8	Mechanism of membrane fouling (Iorhemen <i>et al.</i> , 2016)	23
2.9	Taxonomy of <i>P. speciosa</i>	28
2.10	Interaction between modified membrane and bacterial cell (Basri, 2012)	29
3.1	Flow chart of the research framework	33
3.2	The synthesis process of bio-AgNPs	34
3.3	Zebra fish	36
3.4	The chemical structure of PSf	38
3.5	Ultrafiltration permeation testing unit schematic diagram, (1) feed tank; (2) control valve; (3) pre-treatment tank (4) osmotic pump; (5) flow meter; (6) pressure gauge; (7) filter holder; (8) beaker for collecting permeate	46
3.6	Basic principle of UV-Vis spectrophotometer	47
3.7	Disk –Diffusion technique	48
3.8	Membrane resistance	49
4.1	(a) Sample of AgNO ₃ solution, (b) <i>P. speciosa</i> leaves extract, and (c) mixture of AgNO ₃ with <i>P. speciosa</i> leaves	52

	extract.	
4.2	Bio-AgNPs powder	52
4.3	UV-vis spectrum of silver nanoparticles	54
4.4	XRD spectrum for silver nanoparticles	56
4.5	The deconvulation	56
4.6	FTIR spectrum for leaves extract of <i>P. speciosa</i>	59
4.7	FTIR spectra <i>P. speciosa</i> and bio-AgNPs using KBr	59
4.8	Raman spectra for bio-AgNPs and synthetic AgNPs	60
4.9	FESEM images for bio-AgNPs (a) at magnification of 50,000 x (b) at magnification of 100,000 x	61
4.10	EDX spectra for bio-AgNPs.	62
4.11	Silver nanoparticles TEM image	63
4.12	Average of particles size distribution.	63
4.13	Capping agent	64
4.14	The diagram for biosynthesis process of AgNPs	65
4.15	The mechanism of bacterial disinfection	66
4.16	The zone of inhibition ring for (a) bio-AgNPs and (b) synthetic AgNPs	66
4.17	Comparison cell wall for gram-positive and gram-negative bacteria	67
4.18	Comparison of mortality for Zebra fish	69
4.19	The product of yield for bio-AgNPs.	70
4.20	Perturbation for maximum bio-AgNPs yield production	72
4.21	3D plot for maximum bio-AgNPs yield production	73
4.22	FTIR spectrum for 1% bio-AgNPs/PSf membrane	75
4.23	FTIR spectrum for 1% bio-AgNPs/PSf membrane	76
4.24	FTIR spectrum for 1% bio-AgNPs/PSf membrane	76
4.25	Possible chemical interaction between AgNPs and PSf	77
4.26	Raman spectrum for bio-AgNPs/PSf and AgNPs/PSf membrane	77
4.27	XRD patterns of 1% bio-AgNPs/PSf, 1% AgNPs/PSf and pristine PSf membranes	79

4.28	EDX for (a) pristine PSf, (b) 1% bio-AgNPs/PSf, and (c) 1% AgNPs/PSf membranes	80
4.29	AFM image for pristine PSf membrane	82
4.30	AFM image for concentration 0.1g (B1), 0.3g (B2), 0.5g (B3), 1.0g (B4) of bio-AgNPs/PSf membranes	84
4.31	AFM image for concentration 0.1g (C1), 0.3g (C2), 0.5g (C3), 1.0g (C4) of AgNPs/PSf membranes	85
4.32	Surface roughness value of bio-AgNPs/PSf and AgNPs/PSf membranes	86
4.33	The three layer of cross section for modified membrane	86
4.34	Cross-section SEM image for pristine PSf membrane	88
4.35	Cross-section SEM images for bio-AgNPs/PSf membranes	88
4.36	Cross-section SEM images for AgNPs/PSf membranes.	90
4.37	Mean pore size for bio-AgNPs/PSf and AgNPs/PSf membranes	92
4.38	Porosity value for bio-AgNPs/PSf and AgNPs/PSf membrane	93
4.39	Water flux permeation	95
4.40	(a) The bio-AgNPs/PSf , (b) AgNPs/PSf membranes after permeation. (c) The humic acid rejection result for both membranes.	95
4.41	Interaction between humic acid and encapsulated bio-AgNPs easily attach to organic pollutant	96
4.42	The storage modulus for bio-AgNPs/PSf membranes	97
4.43	The tangen delta for bio-AgNPs/PSf membranes	98
4.44	The storage modulus for AgNPs/PSf membranes	99
4.45	The comparison of storage modulus for mixed matrix PSf membranes	100
4.46	The concentration of silver leaching for AgNPs for filtered samples	101
4.47	The concentration of silver leaching for AgNPs for immersed samples	102
4.48	Antibacterial activity results for bio-AgNPs/PSf and AgNPs/PSf membranes	105

4.49	Normalized flux for bio-AgNPs/PSf membranes	106
4.50	Absolute flux for bio-AgNPs/PSf membranes	106
4.51	Normalized flux for AgNPs/PSf membranes	107
4.52	Absolute flux for AgNPs/PSf membranes	108
4.53	Membrane resistance for difference concentration of bio-AgNPs	110
4.54	Membrane resistance for difference concentration of AgNPs	110
4.55	Fouling resistance ratio (FRR) value	111



LIST OF ABBREVIATIONS

AFM	Atomic Force Microscopy
AgNPs	Silver nanoparticles
ASTM	American Society for Testing and Materials
A	Membrane Area (m ²)
Å	Angstrom
C _p	Solute Concentration in permeate stream
C _f	Solute Concentration in Feed
FTIR	Fourier transform-infrared
FESEM	Field Emission Scanning electron microscopy
ICP-MS	Inductive Coupled Plasma Mass Spectrometer
HA	Humic Acid
J _{wf}	Total of pure water flux
J _{cp}	Total of filtration
J _c	Total of pure water flux after physical cleaning
J _a	Total of pure water flux after chemical cleaning
J _o	Pure Water Flux of clean Membrane
J _f	Pure Water Flux of Fouled Membrane
J _f /J _o	Normalizes flux (initial flux / final flux)
LMH	L/m ² . h
MD	Membrane Distillation
MF	Microfiltration
NF	Nanofiltration
NMP	N-methyl-2-pyrrolidone
PEG	Polyethylene glycol

PWF	Pure Water Flux
PSf	Polysulfone
Q	Permeate volume (L)
R	Rejection of feed components
R_t	Total resistance
R_m	Membrane resistance
R_a	Resistance due to absorption
R_c	Resistance due to cake layer
R_{cp}	Resistance due to cake polarization
RSM	Response surface methodology
RO	Reverse osmosis
SEM	Scanning electron microscope
UF	Ultrafiltration
Wt. %	Weight Percentage
XRD	X-ray diffraction
Δ_t	Time (h)



CHAPTER 1

INTRODUCTION

1.1 Research background

There are various types of water treatment system for eliminating bacteria in water that have been used around the world. Bacteria are found in nearly everywhere in air, water, soil, animals and food. Some bacteria are useful to us and some bacteria are harmful which can cause disease, sickness or other problems. Human needs for clean water in everyday life had led to development of many techniques to eliminate bacteria, such as activated carbon, activated biological filter and the use of membranes in filtration system (Dong *et al.*, 2012). Bacteria separation using membrane technology is an attractive separation approach due to the advantages offered by this process, i.e. high stability and efficiency, ease of operation, low operating cost and energy (Liu *et al.*, 2011).

A membrane separation technology can separate smaller particle as well as able to avoid dangerous contaminate of viruses, fungi and bacteria. Polysulfone membranes (PSf) are known to have superior characteristics by having excellent chemical and thermal stability. Recently, usage of PSf membranes was widely applied in separation fields.

However the main challenge of water treatment by membrane is the fouling phenomena. This phenomena may shorten the membrane service life and reduce the water flux permeation (Zhang *et al.*, 2016). Several methods and strategies for membrane fouling control have been developed over the last two decades. The most common method to help reduce fouling of membranes is by modifying the surface of the membranes. Chemical cleaning also controls membrane fouling. Another popular method to control fouling of membrane is by adding inorganic nanoparticles such as Zinc Oxide (ZnO), Titanium Oxide (TiO), Copper (Cu), and Silver (Ag) inside the

membrane. These nanoparticles will kill bacterial cells and prevent bacterial attachment to membrane surfaces, which reduce the membrane fouling (McCloskey *et al*, 2010).

The effect of additive towards membrane performance is very important in membrane separation. For example, silver is mixed with the membrane because it can enhance the antibacterial properties. Silver nanoparticles are one of the most important materials with diverse applications. Silver is insoluble in water, which is widely used as an additive in numerous materials and product. It occurs naturally, but most of silver nanoparticles is produced synthetically. In this study, silver was produced by synthesis method using leaf extract of *P. speciosa*. Biosynthesized silver nanoparticles (bio-AgNPs) was added to Polysulfone (PSf) membranes in order to investigate the antibacterial properties of the membranes. The leaf extract was used as reducing and stabilizing agent to synthesize silver nanoparticles at room temperature protocol. *P. speciosa* has high composition of reducing agents such as saponins, flavonoids and terpenoids (Kassim *et al.*, 2011). Besides, its antibacterial properties and also abundantly available in Asian countries (Wonghirundecha *et al.*, 2014). The effect of bio-AgNPs concentration towards membrane characteristic and performance was evaluated using permeability, rejection and antibacterial properties of the membrane.

1.2 Problem statement

Clean and healthy water are essential and important things in our daily life nowadays, since most water supply is contaminated with various unwanted and pollutants materials. Therefore, the used of membrane technology that able to filter various size of inorganic and organic materials including bacteria, is essential to various application of wastewater treatment. The used of polymer membrane is more favourable compared to the other types of materials as this membrane offer better processing technique, easy to handle during fabrication and operation as well as low price. PSf membranes have been known to be used in various industrial processes due to their advantages of high chemical resistance stability and tolerance of a wide range of pH.

However, polymer membrane always exposed to the bacteria growth that absorb onto membrane surface during separation process. Therefore, the addition of antibacterial agent in the membrane formulation offer a good alternative in order to improve antibacterial properties as well as membrane performance (Leberknight *et al.*, 2011). As was reported by Sawada *et al.* (2012), the addition of silver particle is able to eliminate or kill almost 99.99 % of E.coli bacteria. Therefore, the silver nanoparticles will be used in this study as an antibacterial agent to prevent the bacteria growth on membrane. Previous study by Procek *et al.* (2011), which shows that silver is a great antibacterial agent when added into membrane formulation. Basri *et al.*, on her investigation confirmed that, most bacteria died with the application of 1 mg Ag/L AgNPs (Basri *et al.*, 2010). Even the used of most additives are able to improve membrane properties and performance, the fact that this inorganic particle has different structure easily create compatibility issue that caused leaching of particles problem. This leaching of silver definitely toxic to human as well as to the environment and aquatic system (Nowack *et al.*, 2012). Therefore the used of biosynthesis particles that able to provide an organic layer which similar and match well with polymer matrix is a good approach in avoiding the leaching of particles problem. The small inorganic particles encapsulated with organic layer surely can bind strongly with the polymer matrix. In fact the biosynthesis technique are seen currently as a sustainable alternative to replace the synthetic chemical laboratory process.

Therefore, in this current work, the silver nanoparticles was synthesized using leaf extract that functioning as reducing and stabilizing agent. As proven in this work, the polymer mixed matrix membrane that integrated with bio-additive nanoparticles (AgNPs) has shown a good strength and better particles distribution inside the membrane. This strong structure will assist the membrane in resisting failure structure when exposing to the high permeation pressure. Furthermore the used natural sources can reduce usage of chemical that not only toxic to the environmental but also is an expensive materials.

1.3 Research objective

The aim of this research study is to fabricate a mixed matrix membranes incorporated with biological synthesis of AgNPs, characterize and test its performance for membrane separation process. The specific objective of this study as follows:

- i. To assess the effect of *P. speciosa* extracts in synthesizing bio-AgNPs particles towards characteristic, properties and performance of the nano synthesized particles.
- ii. To determine the interface structure and bonding between bio-AgNPs and matrix of PSF membrane
- iii. To optimise the operating parameter in synthesizing AgNPs using bio-synthesized technique.
- iv. To evaluate the characteristic, properties and performance of the polymer mixed matric membrane integrated with bio-AgNPs.

1.4 Research scope

In order to achieve the objective, the following scopes are outlined:-

- i. Synthesize bio-AgNPs using leaves extract of *P. speciosa* as reducing and capping agent.
- ii. Characterizing bio-AgNPs via UV-Vis spectroscopy (UV-Vis), Field Emission Scanning Electron Microscope (FESEM), Transmission Electron Microscopy (TEM), X-ray Diffraction (XRD), and Fourier Transform Infrared Spectroscopy (FTIR)
- iii. Optimizing process conditions for AgNPs preparation using Response Surface Methodology (RSM) for different parameters which are weight of *P. speciosa* leaves, incubation time, and temperature.
- iv. Investigating toxicity level of AgNPs using Zebra fish (*Danio rerio*) test.
- v. Preparing polysulfone membrane blended with both silver nanoparticles (bio-AgNPs and AgNPs) and polyethylene glycol (PEG) as polymer additives.

- vi. Investigating the effect of the studied parameters on the membrane performances with respect to the pure water flux and rejection of natural organic matter (NOM).
- vii. Characterizing membrane via Scanning Electron Microscope (SEM), Atomic Force Microscopy (AFM), and X-ray Diffraction (XRD), Fourier Transform Infrared Spectroscopy (FTIR), and Porosity.
- viii. Antibacterial activity test of the prepared membranes by agar diffusion method against *S.aureus* and *E. coli* bacteria.
- ix. Investigating interaction between AgNPs with PSf membrane using Dynamic Mechanical Analysis and Inductive Coupled Plasma Mass Spectrometer.

1.5 Significant and novelty of research

In this work, the biosynthesized particle from extract plant that not only generate antibacterial properties, but also able to provide better interface area as it is coated with an organic compound. This organic compound has the potential to minimize or reduce the incompatibility problem between organic polymer membrane structures that embedded with inorganic particle. The use of biosynthesized particle coated with organic compound provides better interface area for binding mechanism.

Even though, some previous studies have reported the used of Ag substances as an antibacterial agent, however, there is still a lack of research that have focused on bio-AgNPs. Thus, an investigation on the use of leaves extract from *P. speciosa* as reducing agent for biosynthesis of AgNPs in membrane structure has not yet been reported on other previous studies. Even though, the research on nanoparticle has gained a lot of attention nowadays, but this inorganic additive has a weak interface problem, creating the possibility of leaching. The used of biosynthesis not only overcome the incompatible problem but the ability of natural plant extract to coat the synthesized particles with the tailored and desired characteristic that possessed by this plant. Hence, this research is highly significant towards investigating the biosynthesis of AgNPs using an aqueous leaves extract of *P. speciosa* that consisted highly chemical constituents of flavonoids and terpenoids which used as reducing agents. This biosynthesis also can generate finer particles that also can create

homogenous distribution across the matrix membrane that minimize or avoid the membrane failure due weak bonding at interface structure. Last but not least, this approach able to reduce the use of toxic chemical which offer better technique in creating greener technology.



CHAPTER 2

LITERATURE REVIEW

This chapter discussed previous related works on membrane technology and membrane improvement that focus on the mixed matrix membrane. The introduction of some additive in modifying membrane properties and performance was also included. Furthermore, the nanoparticles technology and biosynthesis of nanoparticles using plant extracts was further explored. Lastly, the membrane fabrication and modification has also been discussed.

2.1 Introduction to membrane technology

The interface between two phases acting as a selective barrier is referred to as membrane (Tolaymat *et al.*, 2010). In other words, membranes form a barrier for some substances and allow others to pass through. The membrane filtration process could be a pressure or vacuum-driven and uses a layer of porous material to eliminate unnecessary particles from a suspension. Figure 2.1 shows the basic principle of membrane separation. Membrane separation processes consumed less energy when compared to other methods due to the non-heating operation process.

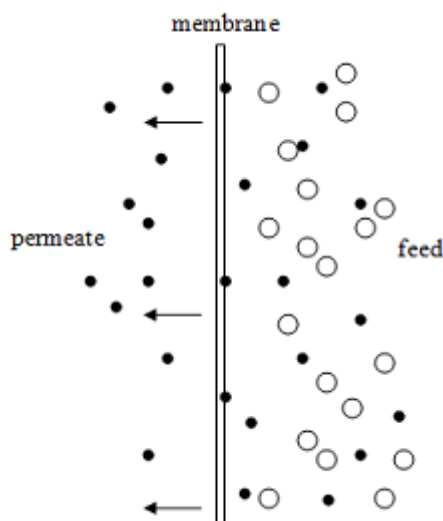


Figure 1.1: Basic principle of membrane separation.

The pressure driven operations involves four types of filtration process which are microfiltration, ultrafiltration, nanofiltration and reverse osmosis. The microfiltration can remove all contaminant particles from fluids or gases in the size range of 0.1 to 10.0 μm (1000 to 10,000 \AA). Suspended solid and bacteria such as starch, *staphylococcus* bacteria and *pseudomonas diminuta* are some examples of filtrate substances treated by microfiltration process (Basri *et al.*, 2010). The microfiltration process involves two techniques, such as, (i) Cross-flow filtration, which is filtered tangentially across the membrane by the flow of the fluid and (ii) Dead-end filtration which is the most widely used. The commonly used operating pressure is in the range 0.02-0.5 MPa. While, the substances size within 100 to 1000 \AA are commonly being removed from a suspension by ultrafiltration with operating pressure of about 0.3 MPa. The viruses, proteins and small colloidal natural organic matter were the common removal substances for ultrafiltration. The similarity between ultrafiltration and microfiltration process can be seen from their operation at low pressure, which is less than 200 kPa (Li *et al.*, 2013).

The nanofiltration pore size was smaller than microfiltration and ultrafiltration. The pore size from 1-10 \AA is capable of removing multivalent ion such as calcium ion (Ca^{2+}) and magnesium ion (Mg^{2+}). Typical operating pressure for nanofiltration is higher compared to microfiltration and ultrafiltration, which is lying between 0.3-4.0 MPa. The nanofiltration membranes are composite membranes consisting of a carrier structure as well as a thin selective layer. Membranes with

REFERENCES

- Abou El-Nour, K. M. M., Eftaiha, A., Al-Warthan, A., & Ammar, R. A. A. (2010). Synthesis and applications of silver nanoparticles. *Arabian Journal of Chemistry*, 3(3), 135–140.
- Ahmed, M. J., Murtaza, G., Mehmood, A., & Bhatti, T. M. (2015). Green synthesis of silver nanoparticles using leaves extract of *Skimmia laureola*: Characterization and antibacterial activity. *Materials Letters*, 153, 10–13.
- Ahmed, S., Ahmad, M., Swami, B. L., & Ikram, S. (2016). A review on plants extract mediated synthesis of silver nanoparticles for antimicrobial applications: A green expertise. *Journal of Advanced Research*, 7(1), 17–28.
- Abdullin, T. I., Bondar, O. V., Shtyrlin, Y. G., Kahraman, M., & Culha, M. (2009). Pluronic block copolymer-mediated interactions of organic compounds with noble metal nanoparticles for SERS analysis. *Langmuir*, 26(7), 5153–5159.
- Alhoshan, M., Alam, J., Dass, L. A., & Al-Homaidi, N. (2013). Fabrication of polysulfone/ZnO membrane: influence of ZnO nanoparticles on membrane characteristics. *Advances in Polymer Technology*, 32(4).
- Alias, S. S., & Mohamad, A. A. (2014). Synthesis of zinc oxide by sol-gel method for photoelectrochemical cells (pp. 41-50). New York, NY: Springer.
- Alias, S. S., Ismail, A. B., & Mohamad, A. A. (2010). Effect of pH on ZnO nanoparticle properties synthesized by sol-gel centrifugation. *Journal of Alloys and Compounds*, 499(2), 231–237.
- Andrade, F. A. C., de Oliveira Vercik, L. C., Monteiro, F. J., & da Silva Rigo, E. C. (2016). Preparation, characterization and antibacterial properties of silver nanoparticles–hydroxyapatite composites by a simple and eco-friendly method. *Ceramics International*, 42(2), 2271–2280.

- Ang, L. Y., Lim, M. E., Ong, L. C., & Zhang, Y. (2011). Applications of upconversion nanoparticles in imaging, detection and therapy. *Nanomedicine*, 6(7), 1273-1288.
- Artale, M. A., Augugliaro, V., Drioli, E., Golemme, G., Grande, C., Loddo, & Schiavello, M. (2001). Preparation and characterisation of membranes with entrapped TiO₂ and preliminary photocatalytic tests. *Annali di chimica*, 91(3-4), 127-136.
- Arthanareeswaran, G., Devi, T. S., & Mohan, D. (2009). Development, characterization and separation performance of organic-inorganic membranes: part II. Effect of additives. *Separation and Purification Technology*, 67(3), 271-281.
- Basri, H., Ismail, A. F., & Aziz, M. (2011). Polyethersulfone (PES)-silver composite UF membrane: effect of silver loading and PVP molecular weight on membrane morphology and antibacterial activity. *Desalination*, 273(1), 72-80.
- Baghizadeh, A., Ranjbar, S., Gupta, V. K., Asif, M., Pourseyedi, S., Karimi, M. J., & Mohammadinejad, R. (2015). Green synthesis of silver nanoparticles using seed extract of *Calendula officinalis* in liquid phase. *Journal of Molecular Liquids*, 207, 159-163.
- Banerjee, P., Satapathy, M., Mukhopahayay, A., & Das, P. (2014). Leaf extract mediated green synthesis of silver nanoparticles from widely available Indian plants: synthesis, characterization, antimicrobial property and toxicity analysis. *Bioresources and Bioprocessing*, 1(1), 3.
- Bar, H., Bhui, D. K., Sahoo, G. P., Sarkar, P., De, S. P., & Misra, A. (2009). Green synthesis of silver nanoparticles using latex of *Jatropha curcas*. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 339(1-3), 134-139.
- Basri, H., Ismail, A. F., Aziz, M., Nagai, K., Matsuura, T., Abdullah, M. S., & Ng, B. C. (2010). Silver-filled polyethersulfone membranes for antibacterial applications - effect of PVP and TAP addition on silver dispersion. *Desalination*, 261(3), 264-271.

- Begum, N. A., Mondal, S., Basu, S., Laskar, R. A., & Mandal, D. (2009). Biogenic synthesis of Au and Ag nanoparticles using aqueous solutions of Black Tea leaf extracts. *Colloids and Surfaces B: Biointerfaces*, 71(1), 113–118.
- Chowdhury, S., Yusof, F., Omer, M., & Sulaiman, N. (2016). Process Optimization of Silver Nanoparticle Synthesis using Response Surface Methodology. *Procedia Engineering*, 148, 992–999.
- Dechsakulthorn, F., Hayes, A., Bakand, S., Joeng, L., & Winder, C. (2008). In vitro cytotoxicity assessment of selected nanoparticles using human skin fibroblasts.
- Dong, C., He, G., Li, H., Zhao, R., Han, Y., & Deng, Y. (2012). Antifouling enhancement of poly(vinylidene fluoride) microfiltration membrane by adding Mg(OH)₂nanoparticles. *Journal of Membrane Science*, 387–388(1), 40–47.
- Dubey, S. P., Lahtinen, M., & Sillanpää, M. (2010). Green synthesis and characterizations of silver and gold nanoparticles using leaf extract of *Rosa rugosa*. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 364(1–3), 34–41.
- Duval, D. J., McCoy, B. J., Risbud, S. H., & Munir, Z. A. (1998). Size selected silicon particles in sol-gel glass by centrifugal processing. *Journal of Applied Physics*, 83(4), 2301-2307.
- Emadzadeh, D., Lau, W. J., Matsuura, T., Rahbari-Sisakht, M., & Ismail, A. F. (2014). A novel thin film composite forward osmosis membrane prepared from PSf–TiO₂ nanocomposite substrate for water desalination. *Chemical Engineering Journal*, 237, 70-80.
- El-gendi, A., Abdalla, H., & Ali, S. (2012). Construction of Ternary Phase Diagram and Membrane Morphology Evaluation for Polyamide / Formic acid / Water System, 6(5), 62–68.
- Fatimah, I. (2016). Green synthesis of silver nanoparticles using extract of *Parkia speciosa* Hassk pods assisted by microwave irradiation. *Journal of Advanced*

Research, 7(6), 961–969.

Fatimah, I. (2016). Green synthesis of silver nanoparticles using extract of *Parkia speciosa* Hassk pods assisted by microwave irradiation. *Journal of Advanced Research*, 7(6), 961-969.

Filimon, A., Stoica, I., Onofrei, M. D., Barga, A., & Dunca, S. (2018). Quaternized polysulfones-based blends: Surface properties and performance in life quality and environmental applications. *Polymer Testing*, 71, 285-295.

Garg, A., Garg, J., Upadhyay, G. C., Agarwal, A., & Bhattacharjee, A. (2015). Evaluation of the Rapidec Carba NP test kit for detection of carbapenemase-producing Gram-negative bacteria. *Antimicrobial Agents and Chemotherapy*, 59(12), 7870-7872.

Gengan, R. M., Anand, K., Phulukdaree, A., & Chuturgoon, A. (2013). A549 lung cell line activity of biosynthesized silver nanoparticles using *Albizia adianthifolia* leaf. *Colloids and Surfaces B: Biointerfaces*, 105, 87-91.

Gehrke, I., & Somborn-schulz, A. (2015). Innovations in nanotechnology for water treatment, 1–17.

Guzmán, M. G., Dille, J., & Godet, S. (2009). Synthesis of silver nanoparticles by chemical reduction method and their antibacterial activity, 104–111.

Hamid, N. A. A., Ismail, A. F., Matsuura, T., Zularisam, A. W., Lau, W. J., Yuliwati, E., & Abdullah, M. S. (2011). Morphological and separation performance study of polysulfone/titanium dioxide (PSF/TiO₂) ultrafiltration membranes for humic acid removal. *Desalination*, 273(1), 85-92.

Harun, Z., Yunus, M. Z., & Nazri, K. (2016). Optimization and characterization of polysulfone membranes made of zinc oxide, polyethylene glycol and eugenol as additives, *Journal of Engineering Science and Technology* 11(7), 1001–1015.

Hettiarachchi, M. A., & Wickramarachchi, P. A. S. R. (2011). Synthesis of chitosan stabilized silver nanoparticles using gamma ray irradiation and characterization.

- Huang, L., Zhao, S., Wang, Z., Wu, J., Wang, J., & Wang, S. (2016). In situ immobilization of silver nanoparticles for improving permeability, antifouling and anti-bacterial properties of ultrafiltration membrane. *Journal of Membrane Science*, 499, 269–281.
- Hong, J., & He, Y. (2012). Effects of nano sized zinc oxide on the performance of PVDF microfiltration membranes. *Desalination*, 302, 71-79.
- Hsiao, I. L., & Huang, Y. J. (2011). Effects of various physicochemical characteristics on the toxicities of ZnO and TiO₂ nanoparticles toward human lung epithelial cells. *Science of the Total Environment*, 409(7), 1219-1228.
- Iorhemen, O., Hamza, R., & Tay, J. (2016). Membrane bioreactor (MBR) technology for wastewater treatment and reclamation: membrane fouling. *Membranes*, 6(2), 33.
- Irfan, M., Moniruzzaman, M., Ahmad, T., Mandal, P. C., Bhattacharjee, S., & Abdullah, B. (2017). Ionic liquid based extraction of flavonoids from *Elaeis guineensis* leaves and their applications for gold nanoparticles synthesis. *Journal of Molecular Liquids*, 241, 270-278.
- Jamaluddin, F., Mohamed, S., & Lajisb, N. (1995). Hypoglycaemic effect of Stigmast-4-en-3-one, from *Parkia speciosa* empty pods, 54, 0–4.
- Jannoo, K., Teerapatsakul, C., Punyanut, A., & Pasanphan, W. (2015). Electron beam assisted synthesis of silver nanoparticle in chitosan stabilizer: Preparation, stability and inhibition of building fungi studies. *Radiation Physics and Chemistry*, 112, 177–188.
- Jeeva, K., Thiagarajan, M., Elangovan, V., Geetha, N., & Venkatachalam, P. (2014). *Caesalpinia coriaria* leaf extracts mediated biosynthesis of metallic silver nanoparticles and their antibacterial activity against clinically isolated pathogens. *Industrial Crops and Products*, 52, 714–720.
- Jucker, C., & Clark, M. M. (1994). Adsorption of aquatic humic substances on hydrophobic ultrafiltration membranes. *Journal of Membrane Science*, 97, 37-52.

- Jurasekova, Z., Marconi, G., Sanchez-Cortes, S., & Torreggiani, A. (2009). Spectroscopic and molecular modeling studies on the binding of the flavonoid luteolin and human serum albumin. *Biopolymers: Original Research on Biomolecules*, 91(11), 917-927.
- Kamisah, Y., Othman, F., Qodriyah, H. M. S., & Jaarin, K. (2013). *Parkia speciosa* hassk.: A potential phytomedicine. *Evidence-Based Complementary and Alternative Medicine*, 2013.
- Kapoor, S., Lawless, D., Kennepohl, P., Meisel, D., & Serpone, N. (1994). Reduction and aggregation of silver ions in aqueous gelatin solutions. *Langmuir*, 10(9), 3018-3022.
- Kassim, J., & Karim, S. A. (2011). Preliminary Studies on Phytochemical Screening of Ulam and Fruit from Malaysia, 8.
- Khan, M. A., Khan, T., & Nadhman, A. (2016). Applications of plant terpenoids in the synthesis of colloidal silver nanoparticles. *Advances in Colloid and Interface Science*, 234, 132-141.
- Kim, J., & Mayfield, S. P. (1997). Protein disulfide isomerase as a regulator of chloroplast translational activation. *Science*, 278(5345), 1954-1957.
- Koseoglu-Imer, D. Y., Kose, B., Altinbas, M., & Koyuncu, I. (2013). The production of polysulfone (PS) membrane with silver nanoparticles (AgNP): physical properties, filtration performances, and biofouling resistances of membranes. *Journal of Membrane Science*, 428, 620-628.
- Krishnaraj, C., Jagan, E. G., Rajasekar, S., Selvakumar, P., Kalaichelvan, P. T., & Mohan, N. (2010). Synthesis of silver nanoparticles using *Acalypha indica* leaf extracts and its antibacterial activity against water borne pathogens. *Colloids and Surfaces B: Biointerfaces*, 76(1), 50-56.
- Krishnaraj, C., Harper, S. L., & Yun, S. I. (2016). In Vivo toxicological assessment of biologically synthesized silver nanoparticles in adult Zebrafish (*Danio rerio*). *Journal of Hazardous Materials*, 301, 480-491.

- Kumar, R., & Ismail, A. F. (2015). Fouling control on microfiltration/ultrafiltration membranes: Effects of morphology, hydrophilicity, and charge. *Journal of Applied Polymer Science*, 132(21).
- Kumar, R., Isloor, A. M., Ismail, A. F., Rashid, S. A., & Ahmed, A. Al. (2013). Permeation, Antifouling and desalination performance of TiO₂nanotube incorporated PSf/CS blend membranes. *Desalination*, 316, 76–84.
- Kuppusamy, P., Ichwan, S. J., Parine, N. R., Yusoff, M. M., Maniam, G. P., & Govindan, N. (2015). Intracellular biosynthesis of Au and Ag nanoparticles using ethanolic extract of Brassica oleracea L. and studies on their physicochemical and biological properties. *Journal of Environmental Sciences*, 29, 151-157.
- Lee, S., Cho, J., & Elimelech, M. (2005). Combined influence of natural organic matter (NOM) and colloidal particles on nanofiltration membrane fouling. *Journal of Membrane Science*, 262(1-2), 27-41.
- Leberknight, J., Wielenga, B., Lee-Jewett, A., & Menkhaus, T. J. (2011). Recovery of high value protein from a corn ethanol process by ultrafiltration and an exploration of the associated membrane fouling. *Journal of Membrane Science*, 366(1-2), 405-412.
- Li, F., Weir, M. D., Chen, J., & Xu, H. H. (2013). Comparison of quaternary ammonium-containing with nano-silver-containing adhesive in antibacterial properties and cytotoxicity. *Dental Materials*, 29(4), 450-461.
- Li, J. F., Xu, Z. L., Yang, H., Feng, C. P., & Shi, J. H. (2008). Hydrophilic microporous PES membranes prepared by PES/PEG/DMAc casting solutions. *Journal of Applied Polymer Science*, 107(6), 4100-4108.
- Liu, J., & Hurt, R. H. (2010). Ion release kinetics and particle persistence in aqueous nano-silver colloids. *Environmental Science & Technology*, 44(6), 2169-2175.
- Li, J. H., Shao, X. S., Zhou, Q., Li, M. Z., & Zhang, Q. Q. (2013). The double effects of silver nanoparticles on the PVDF membrane: Surface hydrophilicity and antifouling performance. *Applied Surface Science*, 265, 663–670.

- Logeswari, P., Silambarasan, S., & Abraham, J. (2015). Synthesis of silver nanoparticles using plants extract and analysis of their antimicrobial property. *Journal of Saudi Chemical Society*, 19(3), 311–317.
- Ma, Y., Shi, F., Ma, J., Wu, M., Zhang, J., & Gao, C. (2011). Effect of PEG additive on the morphology and performance of polysulfone ultrafiltration membranes. *Desalination*, 272(1-3), 51-58.
- Matin, A., Khan, Z., Zaidi, S. M. J., & Boyce, M. C. (2011). Biofouling in reverse osmosis membranes for seawater desalination: phenomena and prevention. *Desalination*, 281, 1-16.
- McShan, D., Ray, P.C., Yu, H., 2014. Molecular toxicity mechanism of nanosilver. *Journal of Food Drug. Anal.* 22, 116–127
- Moghimifar, V., Raisi, A., & Aroujalian, A. (2014). Surface modification of polyethersulfone ultrafiltration membranes by corona plasma-assisted coating TiO₂ nanoparticles. *Journal of Membrane Science*, 461, 69-80.
- Morones, J. R., Elechiguerra, J. L., Camacho, A., Holt, K., Kouri, J. B., Ramírez, J. T., & Yacaman, M. J. (2005). The bactericidal effect of silver nanoparticles. *Nanotechnology*, 16(10), 2346.
- Mulder, J. (2012). *Basic principles of membrane technology*. Springer Science & Business Media.
- Mehrnia, M. R., Mojtahedi, Y. M., & Homayoonfal, M. (2015). What is the concentration threshold of nanoparticles within the membrane structure? A case study of Al₂O₃/PSf nanocomposite membrane. *Desalination*, 372, 75–88.
- Mittal, A. K., Tripathy, D., Choudhary, A., Aili, P. K., Chatterjee, A., Singh, I. P., & Banerjee, U. C. (2015). Bio-synthesis of silver nanoparticles using *Potentilla fulgens* Wall. ex Hook. and its therapeutic evaluation as anticancer and antimicrobial agent. *Materials Science and Engineering C*, 53, 120–127.
- Mohammed, A. E. (2015). Green synthesis, antimicrobial and cytotoxic effects of silver nanoparticles mediated by *Eucalyptus camaldulensis* leaf extract. *Asian Pacific Journal of Tropical Biomedicine*, 5(5), 382–386.

- Mollahosseini, A., Rahimpour, A., Jahamshahi, M., Peyravi, M., & Khavarpour, M. (2012). The effect of silver nanoparticle size on performance and antibacteriability of polysulfone ultrafiltration membrane. *Desalination*, 306, 41–50.
- Nabikhan, A., Kandasamy, K., Raj, A., & Alikunhi, N. M. (2010). Synthesis of antimicrobial silver nanoparticles by callus and leaf extracts from saltmarsh plant, *Sesuvium portulacastrum* L. *Colloids and Surfaces B: Biointerfaces*, 79(2), 488–493.
- Nasrollahzadeh, M., Sajadi, M. M., Babaei, F., & Maham, M. (2015). *Euphorbia helioscopia* Linn as a green source for synthesis of silver nanoparticles and their optical and catalytic properties. *Journal of Colloid and Interface Science*, 450, 374–380.
- Naik, B. R., Gowreeswari, G. S., Singh, Y., Satyavathi, R., Daravath, S. S., & Reddy, P. R. (2014). Bio-synthesis of silver nanoparticles from leaf extract of *Pongamia pinnata* as an effective larvicide on dengue vector *Aedes albopictus* (Skuse)(Diptera: Culicidae). *Advances in Entomology*, 2(02), 93.
- Noruzi, M. (2015). Biosynthesis of gold nanoparticles using plant extracts. *Bioprocess and Biosystems Engineering*, 38(1), 1-14.
- Nowack, B., Ranville, J. F., Diamond, S., Gallego-Urrea, J. A., Metcalfe, C., Rose, & Klaine, S. J. (2012). Potential scenarios for nanomaterial release and subsequent alteration in the environment. *Environmental Toxicology and Chemistry*, 31(1), 50-59.
- Ng, L. Y., Mohammad, A. W., Leo, C. P., & Hilal, N. (2013). Polymeric membranes incorporated with metal/metal oxide nanoparticles: A comprehensive review. *Desalination*, 308, 15–33.
- Ouyang, K., Yu, X. Y., Zhu, Y., Gao, C., Huang, Q., & Cai, P. (2017). Effects of humic acid on the interactions between zinc oxide nanoparticles and bacterial biofilms. *Environmental Pollution*, 231, 1104-1111.
- Panyala, N. R., Peña-Méndez, E. M., & Havel, J. (2008). Silver or silver nanoparticles: a hazardous threat to the environment and human health?. *Journal of Applied Biomedicine (De Gruyter Open)*, 6(3).

- Philip, D. (2010). Green synthesis of gold and silver nanoparticles using *Hibiscus rosa sinensis*. *Physica E: Low-Dimensional Systems and Nanostructures*, 42(5), 1417–1424.
- Qu, F., Liang, H., Zhou, J., Nan, J., Shao, S., Zhang, J., & Li, G. (2014). Ultrafiltration membrane fouling caused by extracellular organic matter (EOM) from *Microcystis aeruginosa*: effects of membrane pore size and surface hydrophobicity. *Journal of membrane science*, 449, 58-66.
- Ramachandran, R., Krishnaraj, C., Sivakumar, A. S., Prasannakumar, P., Kumar, V. A., Shim, K. S., ... & Yun, S. I. (2017). Anticancer activity of biologically synthesized silver and gold nanoparticles on mouse myoblast cancer cells and their toxicity against embryonic zebrafish. *Materials Science and Engineering: C*, 73, 674-683.
- Ravindran, A., Chandran, P., & Khan, S. S. (2013). Biofunctionalized silver nanoparticles: advances and prospects. *Colloids and Surfaces B: Biointerfaces*, 105, 342-352.
- Ribeiro, F., Gallego-Urrea, J. A., Jurkschat, K., Crossley, A., Hassellöv, M., Taylor, & Loureiro, S. (2014). Silver nanoparticles and silver nitrate induce high toxicity to *Pseudokirchneriella subcapitata*, *Daphnia magna* and *Danio rerio*. *Science of the Total Environment*, 466, 232-241.
- Sawada, I., Fachrul, R., Ito, T., Ohmukai, Y., Maruyama, T., & Matsuyama, H. (2012). Development of a hydrophilic polymer membrane containing silver nanoparticles with both organic antifouling and antibacterial properties. *Journal of Membrane Science*, 387–388(1), 1–6.
- Saljoughi, E., Amirilargani, M., & Mohammadi, T. (2010). Effect of PEG additive and coagulation bath temperature on the morphology, permeability and thermal/chemical stability of asymmetric CA membranes. *Desalination*, 262(1-3), 72-78.
- Shankar, P. D., Shobana, S., Karuppusamy, I., Pugazhendhi, A., Ramkumar, V. S., Arvindnarayan, S., & Kumar, G. (2016). A review on the biosynthesis of metallic nanoparticles (gold and silver) using bio-components of microalgae:

Formation mechanism and applications. *Enzyme and Microbial Technology*, 95, 28-44.

Shen, J. N., Ruan, H. M., Wu, L. G., & Gao, C. J. (2011). Preparation and characterization of PES–SiO₂ organic–inorganic composite ultrafiltration membrane for raw water pretreatment. *Chemical Engineering Journal*, 168(3), 1272-1278.

Sheng, J. (1991). Separation of dichloroethane-trichloroethylene mixtures by means of a membrane pervaporation process. *Desalination*, 80(1), 85-95.

Shi, W., Li, H., Zhou, R., Zhang, H., & Du, Q. (2016). Biodiesel production from soybean oil by quaternized polysulfone alkali-catalyzed membrane. *Bioresource Technology*, 210, 43-48.

Shi, Q., Meng, J. Q., Xu, R. S., Du, X. L., & Zhang, Y. F. (2013). Synthesis of hydrophilic polysulfone membranes having antifouling and boron adsorption properties via blending with an amphiphilic graft glycopolymers. *Journal of Membrane Science*, 444, 50-59.

Teli, S. B., Molina, S., Sotto, A., Calvo, E. G., & Abajob, J. D. (2013). Fouling resistant polysulfone–PANI/TiO₂ ultrafiltration nanocomposite membranes. *Industrial & Engineering Chemistry Research*, 52(27), 9470-9479.

Templeton, A. C., Pietron, J. J., Murray, R. W., & Mulvaney, P. (2000). Solvent refractive index and core charge influences on the surface plasmon absorbance of alkanethiolate monolayer-protected gold clusters. *The Journal of Physical Chemistry B*, 104(3), 564-570.

Theivasanthi, T., & Alagar, M. (2011). Studies of copper nanoparticles effects on micro-organisms. *arXiv preprint arXiv:1110.1372*.

Tolaymat, T. M., El Badawy, A. M., Genaidy, A., Scheckel, K. G., Luxton, T. P., & Suidan, M. (2010). An evidence-based environmental perspective of manufactured silver nanoparticle in syntheses and applications: A systematic review and critical appraisal of peer-reviewed scientific papers. *Science of the*

Total Environment, 408(5), 999–1006.

Van, A. R., Lange, A., Moorhouse, A., Paszkiewicz, K., Ball, K., Johnston, B., & Santos, E. M. (2013). Molecular mechanisms of toxicity of silver nanoparticles in zebrafish embryos. *Environmental Science & Technology*, 47(14), 8005-8014.

Vijayaraghavan, K., & Ashokkumar, T. (2017). Plant-mediated biosynthesis of metallic nanoparticles: a review of literature, factors affecting synthesis, characterization techniques and applications. *Journal of Environmental Chemical Engineering*, 5(5), 4866-4883.

Veerasamy, R., Xin, T. Z., Gunasagaran, S., Xiang, T. F. W., Yang, E. F. C., Jeyakumar, N., & Dhanaraj, S. A. (2011). Biosynthesis of silver nanoparticles using mangosteen leaf extract and evaluation of their antimicrobial activities. *Journal of Saudi Chemical Society*, 15(2), 113–120.

Velusamy, P., Das, J., Pachaiappan, R., Vaseeharan, B., & Pandian, K. (2015). Greener approach for synthesis of antibacterial silver nanoparticles using aqueous solution of neem gum (*Azadirachta indica* L.). *Industrial Crops and Products*, 66(1), 103–109.

Vidhu, V. K., Aromal, S. A., & Philip, D. (2011). Green synthesis of silver nanoparticles using *Macrotyloma uniflorum*. *Spectrochimica Acta - Part A: Molecular and Biomolecular Spectroscopy*, 83(1), 392–397.

Wang, J., Xu, Y., Zhu, L., Li, J., & Zhu, B. (2008). Amphiphilic ABA copolymers used for surface modification of polysulfone membranes, Part 1: Molecular design, synthesis, and characterization. *Polymer*, 49(15), 3256–3264.

Wong, S. W., Leung, P. T., Djurišić, A. B., & Leung, K. M. (2010). Toxicities of nano zinc oxide to five marine organisms: influences of aggregate size and ion solubility. *Analytical and Bioanalytical Chemistry*, 396(2), 609-618.

Wonghirundecha, S., Benjakul, S., & Sumpavapol, P. (2014). Total phenolic content, antioxidant and antimicrobial activities of stink bean (*Parkia speciosa* Hassk.) pod extracts. *Songklanakarin Journal Science Technology*, 36(3), 300-8.

- Xia, G., Liu, T., Wang, Z., Hou, Y., Dong, L., Zhu, J., & Qi, J. (2016). The effect of silver nanoparticles on zebrafish embryonic development and toxicology. *Artificial Cells, Nanomedicine, and Biotechnology*, 44(4), 1116-1121.
- Yan, L., Li, Y. S., Xiang, C. B., & Xianda, S. (2006). Effect of nano-sized Al₂O₃-particle addition on PVDF ultrafiltration membrane performance. *Journal of Membrane Science*, 276(1-2), 162-167.
- Yilmaz, M., Turkdemir, H., Kilic, M. A., Bayram, E., Cicek, A., Mete, A., & Ulug, B. (2011). Biosynthesis of silver nanoparticles using leaves of *Stevia rebaudiana*. *Materials Chemistry and Physics*, 130(3), 1195–1202.
- Yuehuei, H., & An, R. J. F. (2000). Handbook of bacterial adhesion: principles, methods, and applications.
- Yuliwati, E., & Ismail, A. F. (2011). Effect of additives concentration on the surface properties and performance of PVDF ultrafiltration membranes for refinery produced wastewater treatment. *Desalination*, 273(1), 226-234.
- Yunos, M. Z. (2014). Development of polysulfone/Silver oxide membranes for separation of Natural Organic Matters (NOM). Universiti Tun Hussein Onn Malaysia: Ph. D. Thesis.
- Yunos, M. Z., Harun, Z., Basri, H., & Ismail, A. F. (2012). Effects of water as non-solvent additive on performance of polysulfone ultrafiltration membrane. In *Advanced Materials Research* (Vol. 488, pp. 46-50). Trans Tech Publications.
- Zhang, Y., Cheng, X., Zhang, Y., Xue, X., & Fu, Y. (2013). Biosynthesis of silver nanoparticles at room temperature using aqueous aloe leaf extract and antibacterial properties. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 423, 63–68.
- Zhang, J., Gao, D., Yang, G., Zhang, J., Shi, Z., Zhang, & Xue, D. (2011). Synthesis and magnetic properties of Zr doped ZnO Nanoparticles. *Nanoscale Research Letters*, 6(1), 587.

- Zhang, X. Q., Yin, L. H., Meng, T. A. N. G., & PU, Y. P. (2011). ZnO, TiO₂, SiO₂, and Al₂O₃ nanoparticles-induced toxic effects on human fetal lung fibroblasts. *Biomedical and Environmental Sciences*, 24(6), 661-669.
- Zhang, D. Y., Liu, J., Shi, Y. S., Wang, Y., Liu, H. F., Hu, Q. L., & Zhu, J. (2016). Antifouling polyimide membrane with surface-bound silver particles. *Journal of Membrane Science*, 516, 83-93.
- Zodrow, K., Brunet, L., Mahendra, S., Li, D., Zhang, A., Li, Q., & Alvarez, P. J. J. (2009). Polysulfone ultrafiltration membranes impregnated with silver nanoparticles show improved biofouling resistance and virus removal. *Water Research*, 43(3), 715–723.

